

# Product Characterization

Purity vs. Characterization

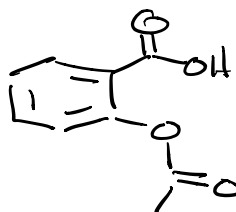
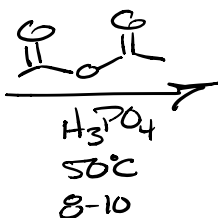
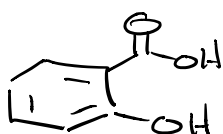
100%  
one  
material

Chromatography  
Side products  
Amounts

mp  
Chemical test  
Spectroscopy

NMR  $^{13}\text{C}$ ,  $^1\text{H}$   
FTIR  
Mass Spec

Chromatography GC-mass spec



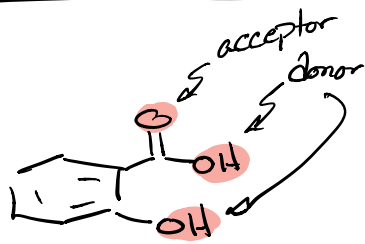
mp

FeCl<sub>3</sub>

FTIR

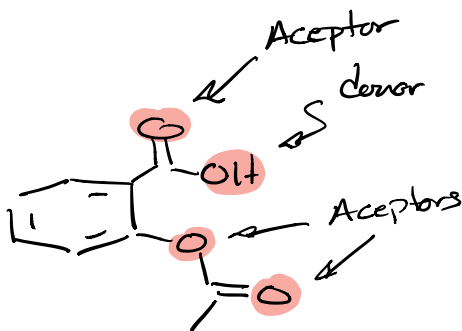
- How to sample a solid
- Spectrum analysis

## Melting Point



Hydrogen bonding

157 - 158 °C



Less H-bonding ↓

more mass ↑

$C_2H_2O$  48 g/mol

134.5 - 135.5 °C

134.5 - 135.5 °C ⇒ product is as expected

128 - 132 °C ⇒ Broad & depressed from Lit value  
⇒ impure & contains starting material

147 - 154 °C ⇒ Broad & depressed from starting material  
⇒ Majority of recovered product is starting material  
acetylsalicylic acid impurity in starting material  
⇒ Run failed

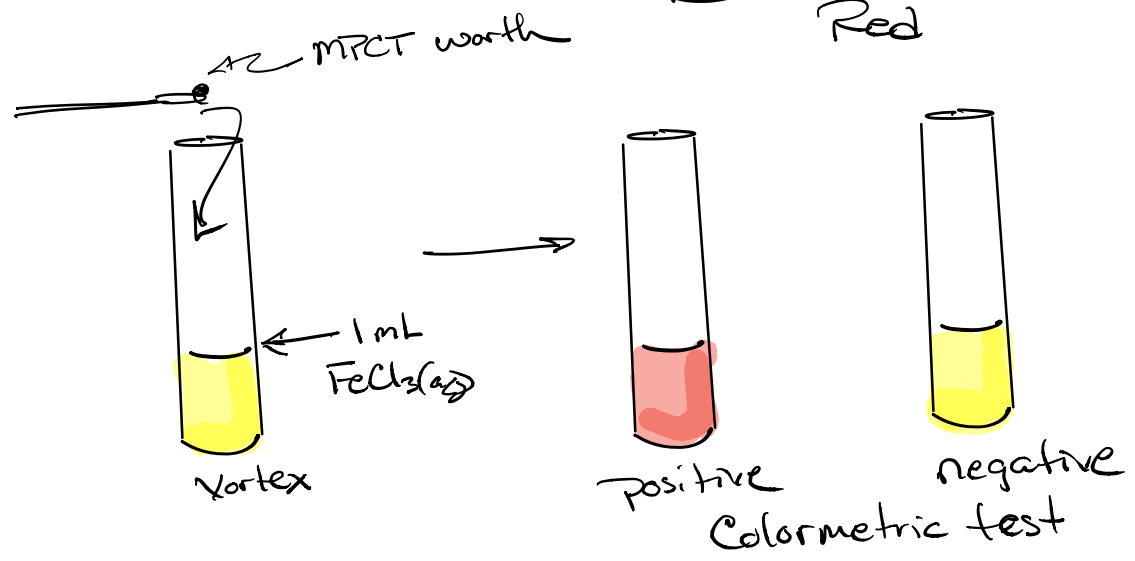
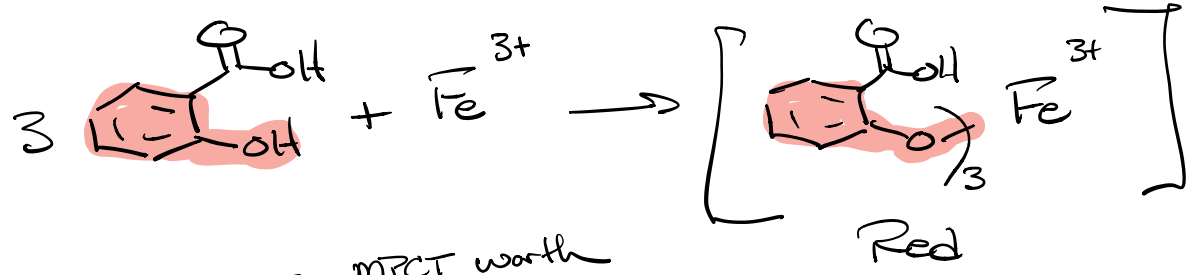
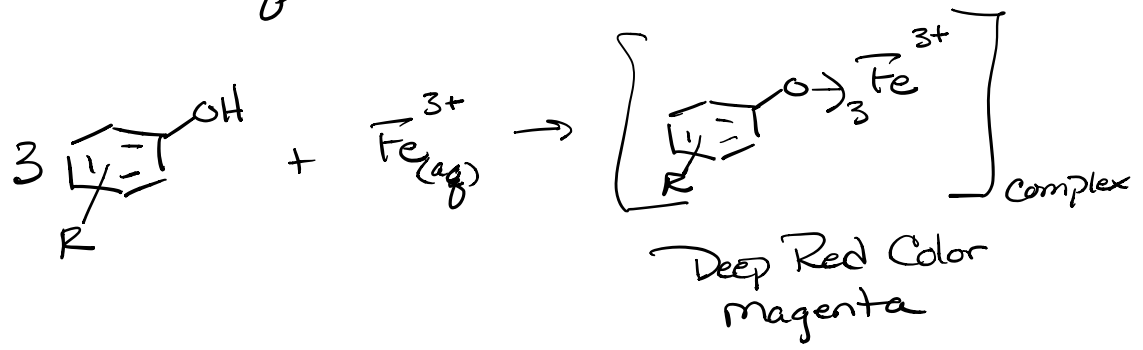
# Ferric Chloride test

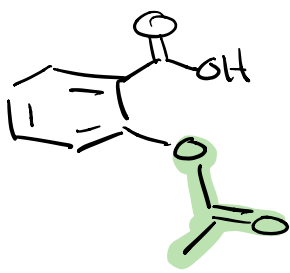
Test for phenol 

Ferric ion  $Fe^{3+}$  (Iron(III))

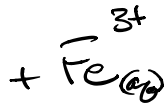
Ferrous ion  $Fe^{2+}$  (Iron(II))

$FeCl_3(aq)$  clear & yellow tinge





no phenol



100



100



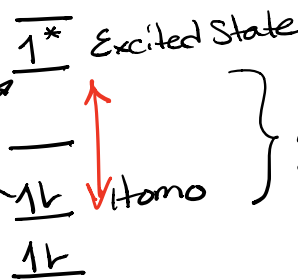
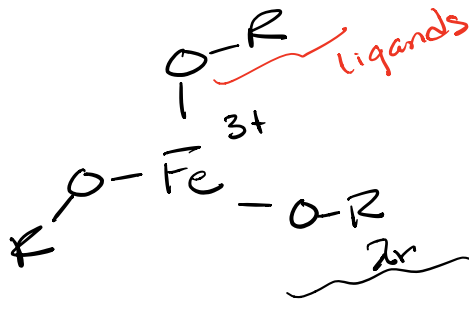
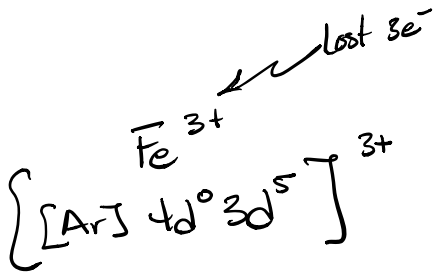
negative  
Consistent w/ expected  
acetylsalicylic  
Acid

134-135°C

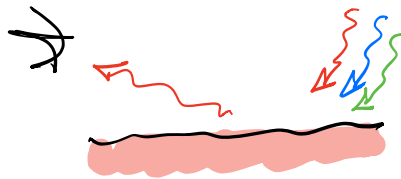
Some  
Phenol  
Contamination  
Some Salicylic acid  
Present

127-132°C

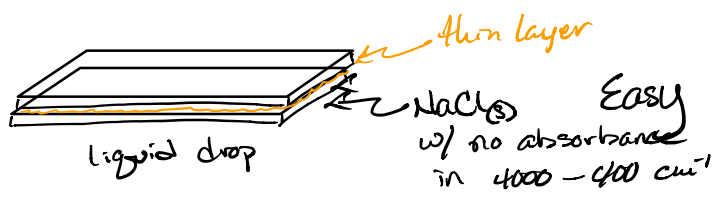
Run failed  
Starting  
Material  
Recov.  
147-152°C



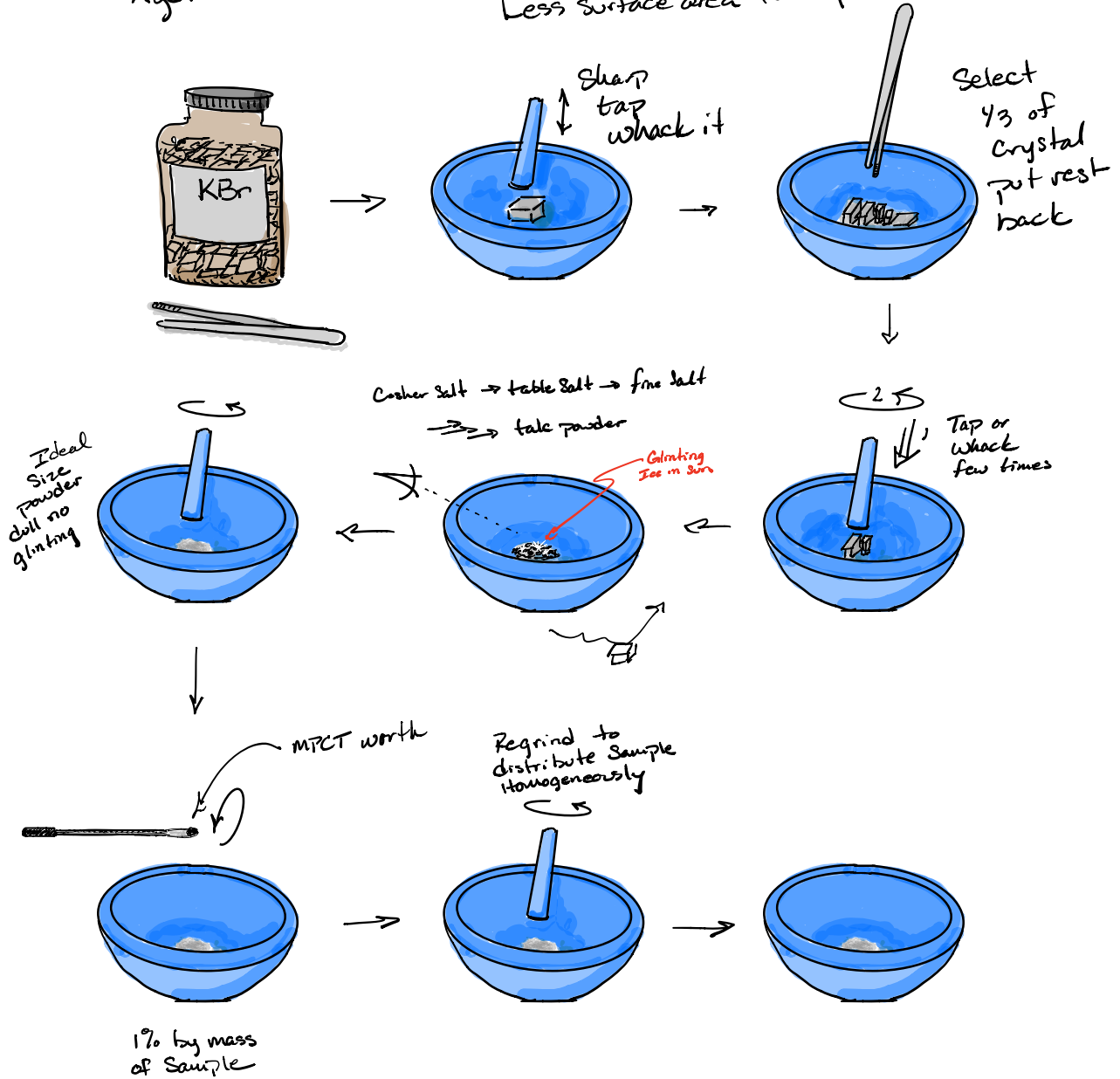
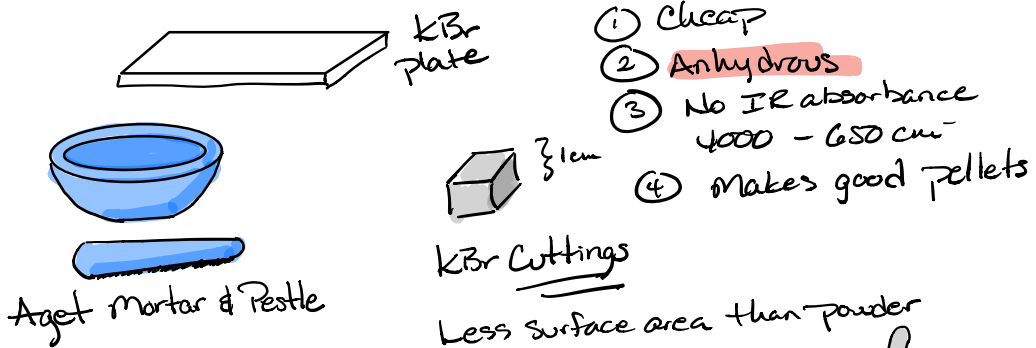
$\Delta E = 2r$   
Energy absorbed

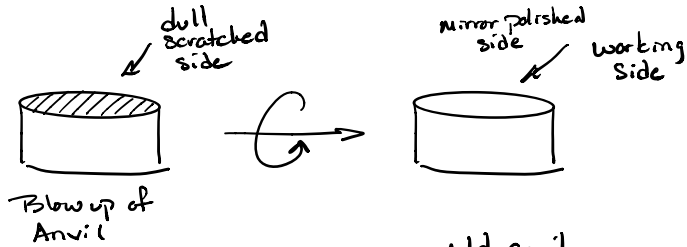
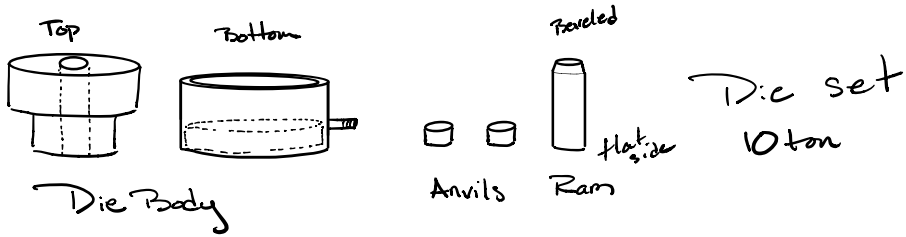


# Solid phase FTIR

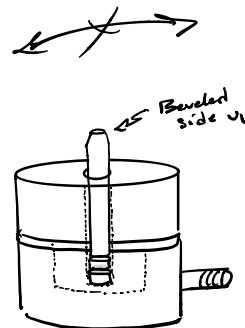
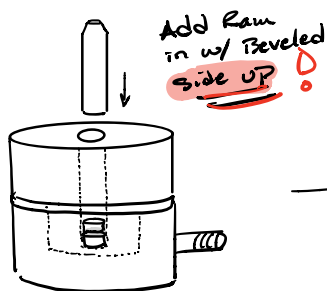
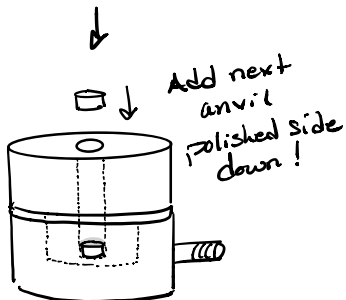
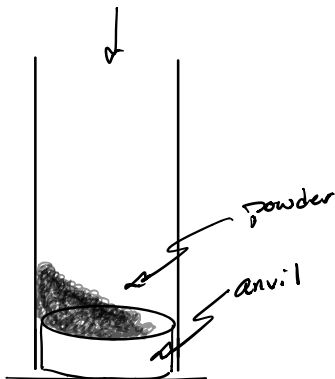
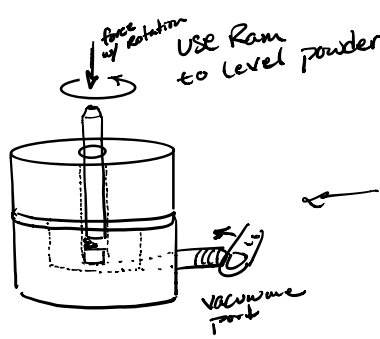
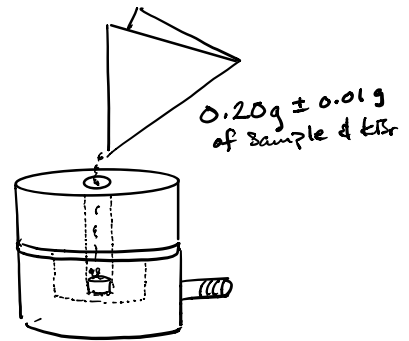
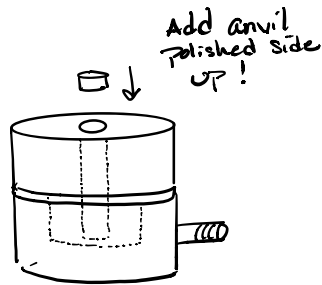
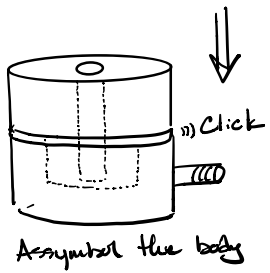


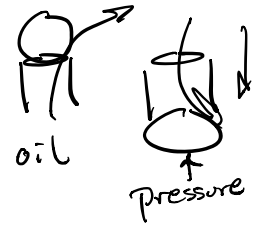
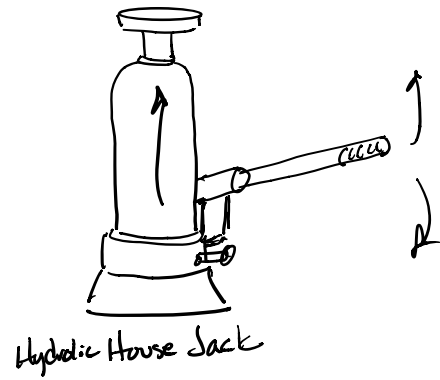
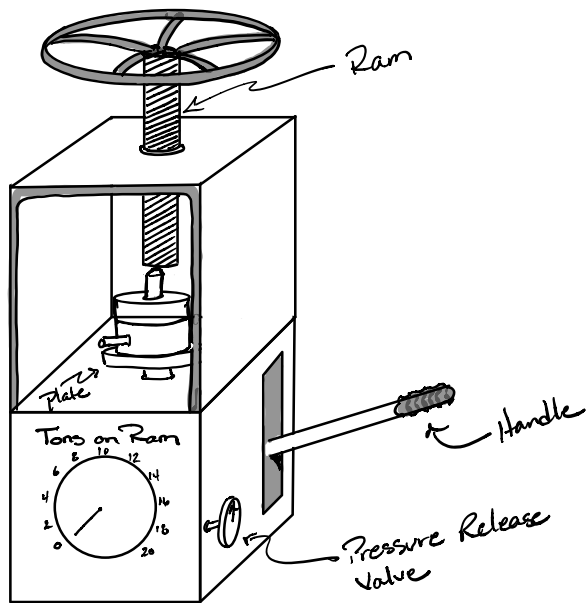
# Solid KBr Pellet





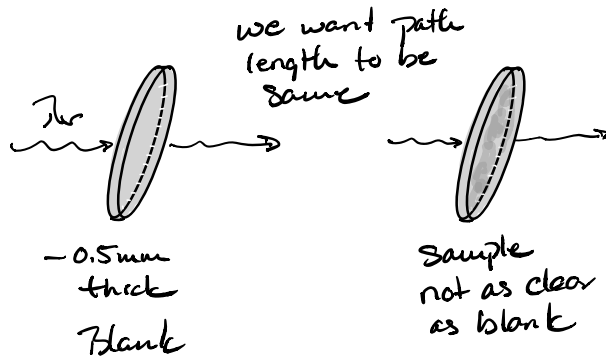
Blank 0.20g





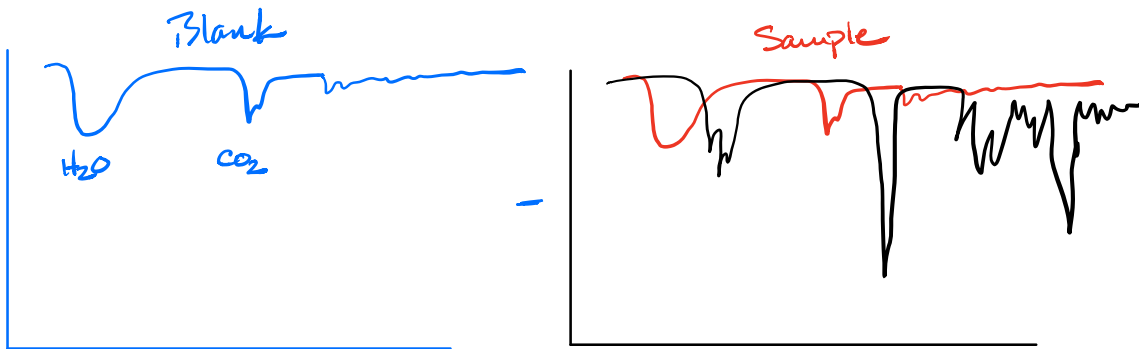
- ① Close pressure valve  
clockwise
- ② Insert & center the die
- ③ Bring ram down to  
contact with die ram
- ④ Crank up to 10 tons  
1-2 min
- ⑤ Release pressure  
 $\frac{1}{4}$  turn counter clockwise  
of valve



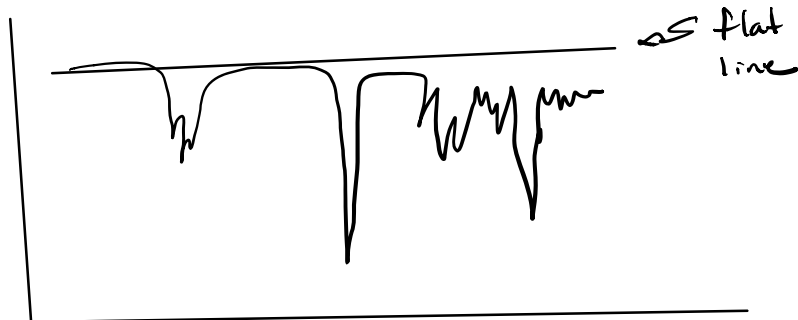


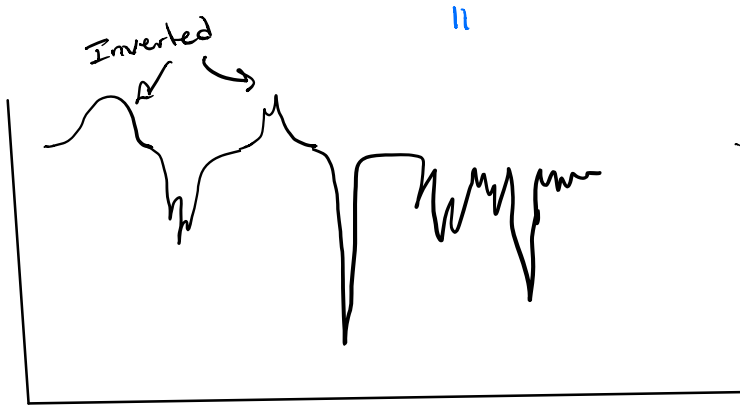
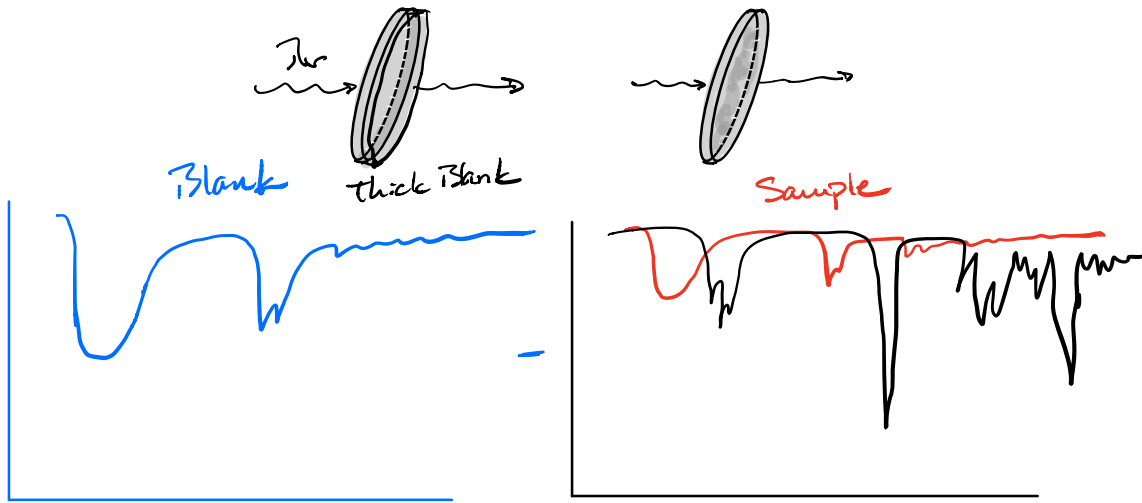
$H_2O$  &  $CO_2$   
 $3400-3200$        $2400$   
 $cm^{-1}$        $cm^{-1}$   
 strong

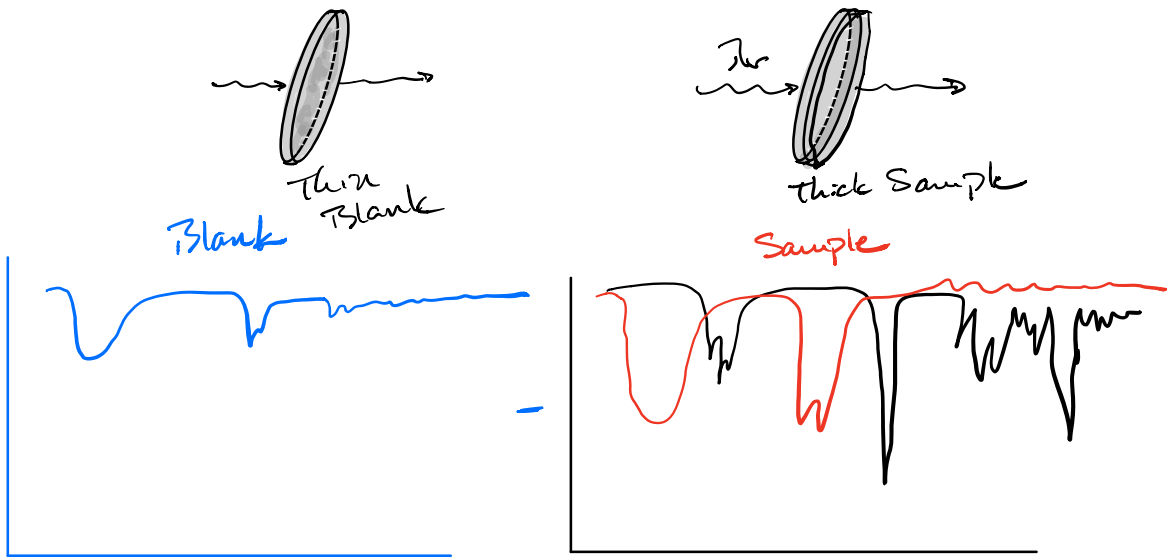
Blank w/ same path length as sample



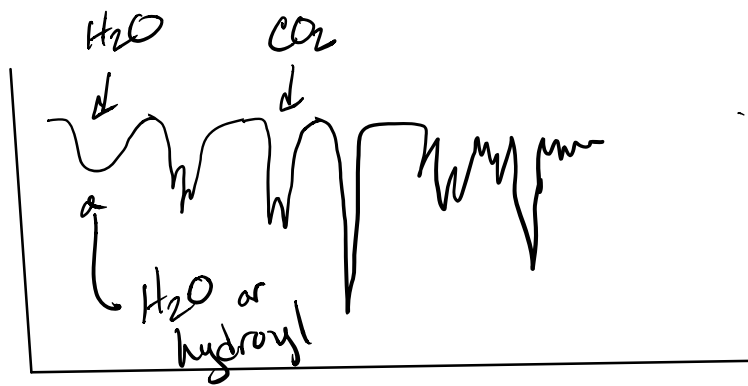
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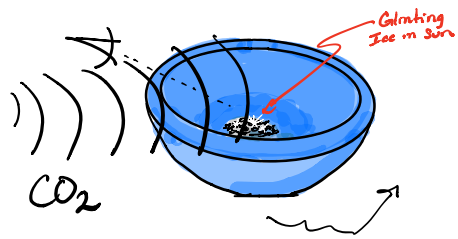






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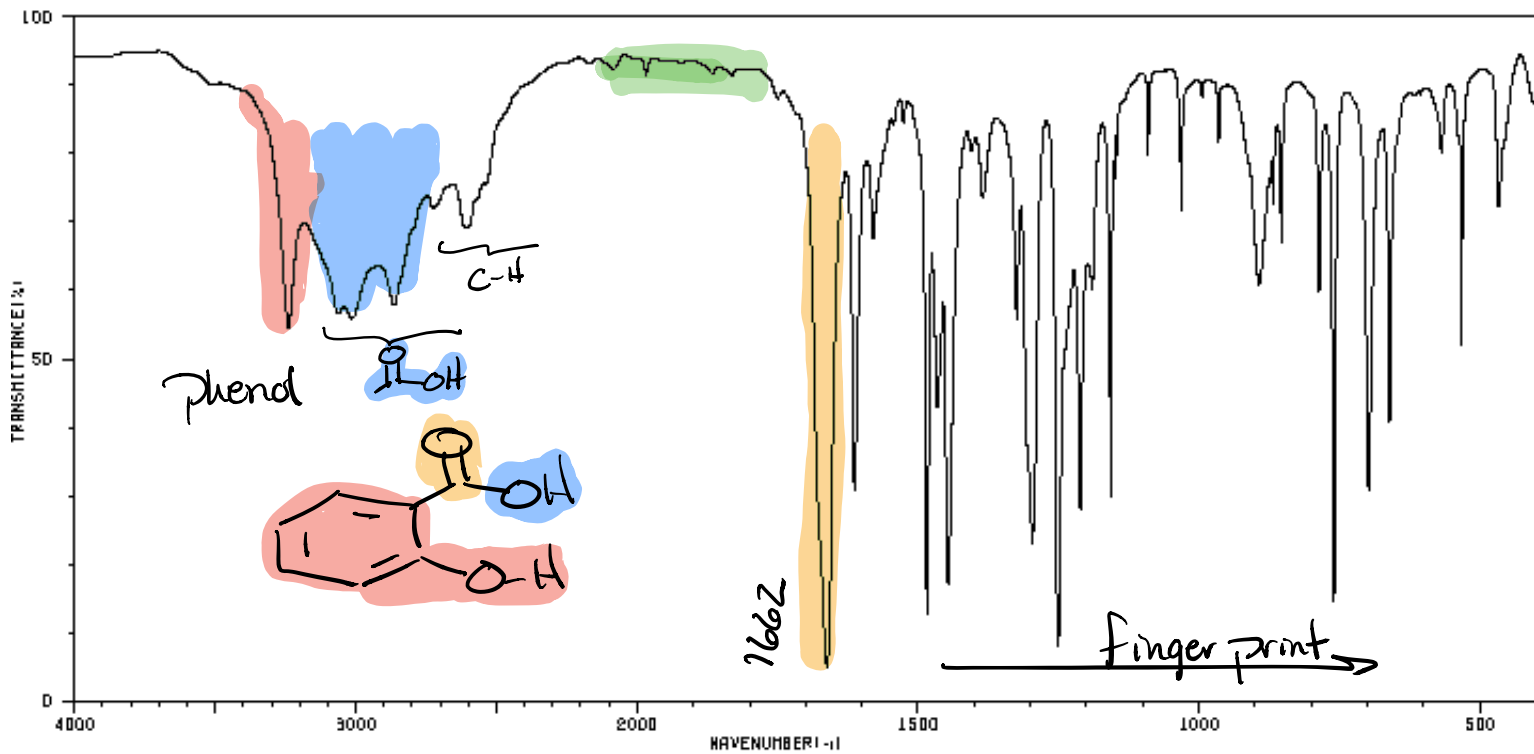




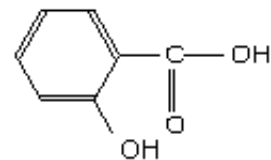
KBr absorbs CO<sub>2</sub>

SALICYLIC ACID

C<sub>7</sub>H<sub>6</sub>O<sub>3</sub>



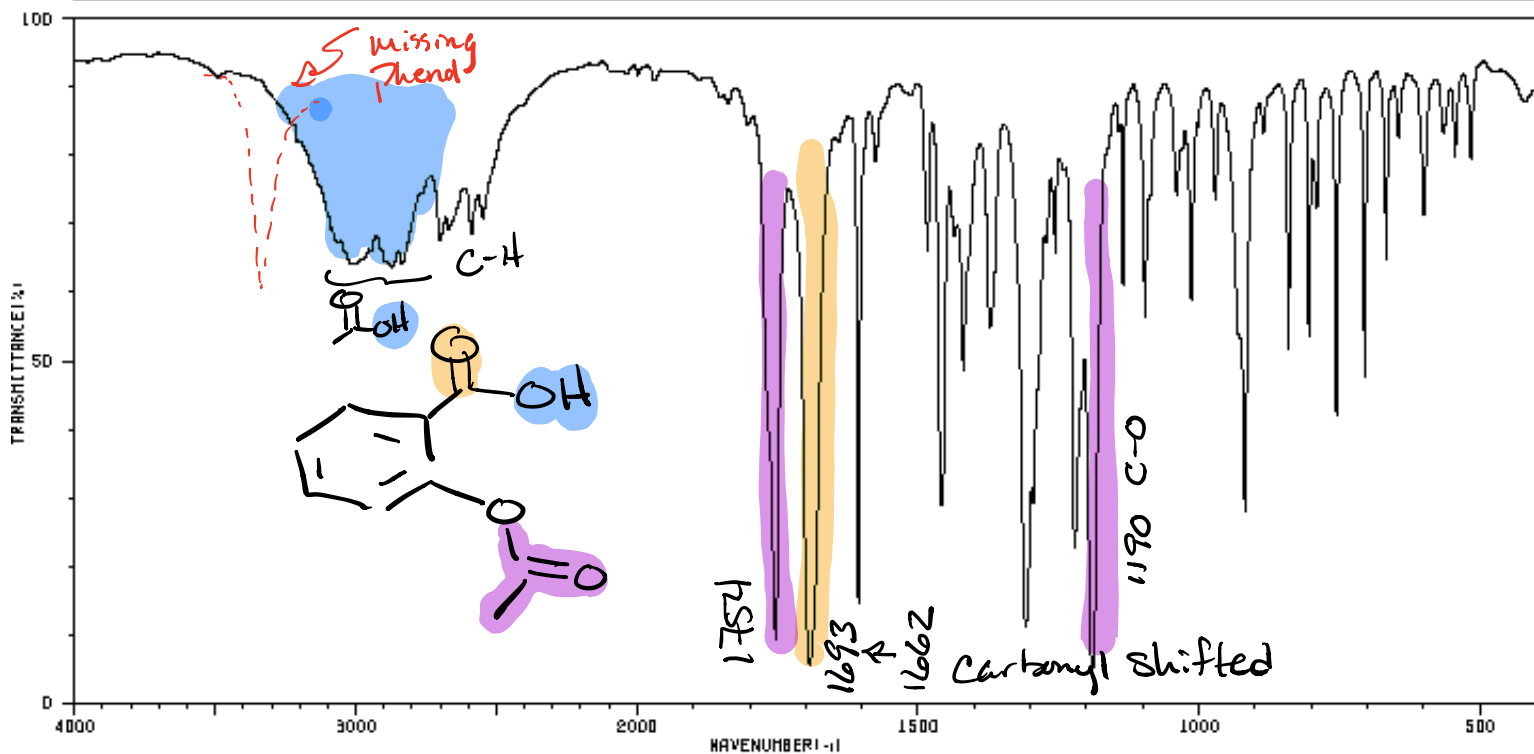
3240	62	1680	64	1326	63	1161	72	786	67
3013	53	1527	81	1297	21	1091	77	760	13
2864	55	1484	12	1251	7	1032	68	699	29
2724	70	1467	41	1239	49	966	79	661	38
2605	66	1447	16	1212	26	893	58	569	77
1662	4	1405	77	1190	58	868	70	533	50
1613	29	1386	70	1167	28	863	64	467	70



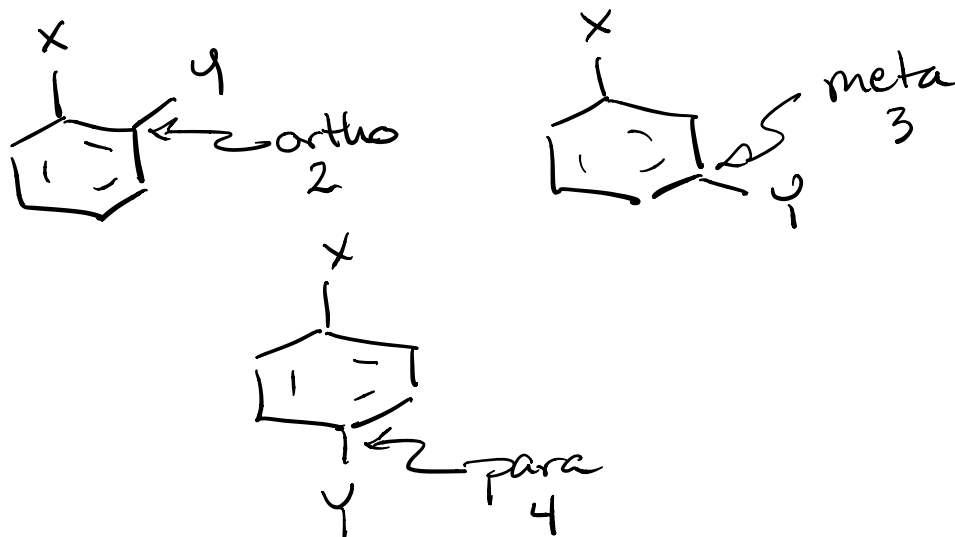
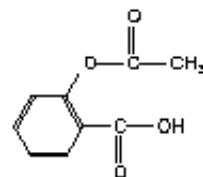
O-ACETOXYBENZOIC ACID

ortho-

C<sub>9</sub>H<sub>8</sub>O<sub>4</sub>



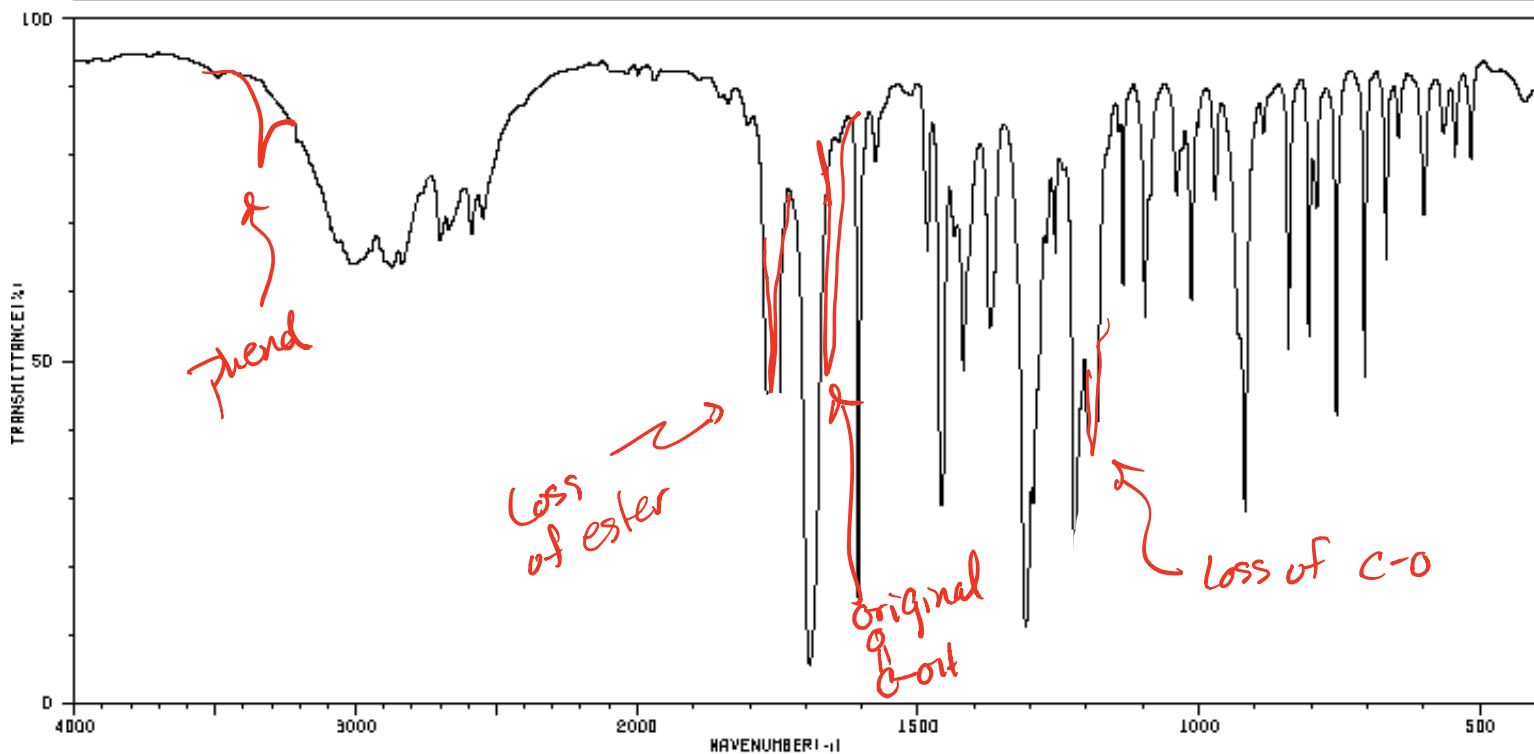
3006	62	2589	66	1436	66	1222	21	918	26
2983	62	2548	68	1420	47	1190	4	841	50
2891	62	1754	9	1372	53	1136	58	805	52
2872	62	1693	6	1308	10	1096	66	766	41
2834	62	1606	14	1295	28	1014	57	706	46
2701	66	1483	64	1272	64	971	70	667	62
2670	66	1469	27	1267	64	928	62	600	68



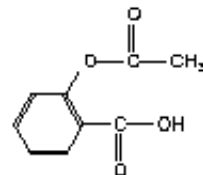
O-ACETOXYBENZOIC ACID

*If Contaminated with starting material*

$C_9H_8O_4$



3006	62	2589	66	1436	66	1222	21	918	26
2983	62	2548	68	1420	47	1190	4	841	50
2891	62	1754	9	1372	53	1136	58	805	52
2872	62	1693	6	1308	10	1096	66	766	41
2834	62	1606	14	1295	28	1014	57	706	46
2701	66	1483	64	1272	64	971	70	667	62
2670	66	1469	27	1267	64	928	62	600	68



- 2 Solubility
  - 3 Crystallizations
  - 4 Extraction
- 

} Lab notes  
Questions from  
Pavia → Question  
Section  
&

Report Section

5 Chromatography nothing due

6 Distillation nothing due

7 IR Spectroscopy nothing due

8 Synthesis of Acetylsalicylic acid \* writing assignment